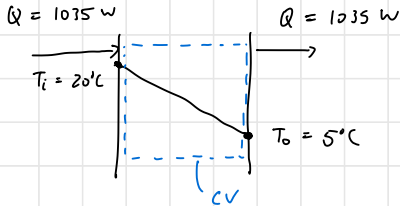


Example

Temperatures at the inner and outer surfaces of a brick wall are 20°C and 5°C , respectively. The rate of heat transfer through the wall is 1035 W .

What is the total rate of entropy generation in the wall?



$$\begin{aligned}\Delta u &= \sum Q - \sum W \\ &= Q_{in} - Q_{out}\end{aligned}$$

$$\frac{ds}{dt} = \dot{S}_{in} - \dot{S}_{out} + \dot{S}_{gen}$$

$$\frac{ds}{dt} = \sum \frac{\dot{Q}_k}{T_k} + \dot{S}_{gen}$$

steady state

$$0 = \frac{1035 [\text{W}]}{293 [\text{K}]} - \frac{1035 [\text{W}]}{278 [\text{K}]} + \dot{S}_{gen}$$

$$\dot{S}_{gen} = 0.191 [\text{W/K}]$$

Entropy generation ≥ 0 \therefore Process is valid
 > 0 \therefore Process has irreversibility

Heat transfer through
a finite temp. difference